

## The "TdS" equations (solid, liquid, or gas)

Recall that we had  $TdS = dU + PdV - \mu dN$   
 If we insist that  $N = \text{const}$ , then we can write

$$TdS = dU + PdV$$

We can manipulate and expand the various functions, rearranging etc (for example see the kind of thing we did in #9 and #14 on the web page) to find:

$$\textcircled{\#1} \quad TdS = C_V dT + T \left. \frac{\partial P}{\partial T} \right|_V dV$$

$$\textcircled{\#2} \quad TdS = C_P dT - T \left. \frac{\partial V}{\partial T} \right|_P dP$$

$$\textcircled{\#3} \quad TdS = C_P \left. \frac{\partial T}{\partial V} \right|_P dV + C_V \left. \frac{\partial T}{\partial P} \right|_V dP$$

} all are  
gas variables  
choose  
depending  
on what  
you  
know

If we continue to substitute:

$$\textcircled{\#1a} \quad TdS = C_V dT + T \left. \frac{\beta}{k} dV \right.$$

$$\textcircled{\#2a} \quad TdS = C_P dT - \beta V T dP$$

$$\textcircled{\#3a} \quad TdS = \frac{C_P}{\beta V} dV + \frac{k}{\beta} C_V dP$$

} choose if  
 $\beta, k$  const.  
over range  
of interest

$$\text{where } \beta = \left. \frac{1}{V} \frac{\partial V}{\partial T} \right|_P \quad \kappa = - \left. \frac{1}{V} \frac{\partial V}{\partial P} \right|_T$$